

**DYSIDAMIDE, A NOVEL HEXACHLORO-METABOLITE FROM A RED SEA
SPONGE DYSIDEA SP**

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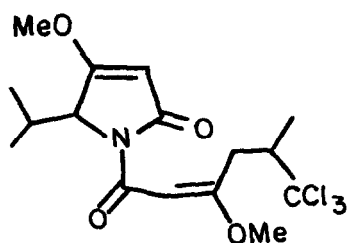
Abstract A hexachloro-metabolite, dysidamide (2) has been
isolated from a Red Sea sponge Dysidea sp and its structure
determined by MS, 1D and 2D NMR spectroscopy

The sponge genus Dysidea has been the source of diverse organic metabolites which have
included bromophenols¹, sesquiterpenoids² and chlorinated compounds³ exemplified by (1)^{3a}
We report the isolation of a hexachloro-metabolite (2), for which we propose the name
dysidamide, from Dysidea sp

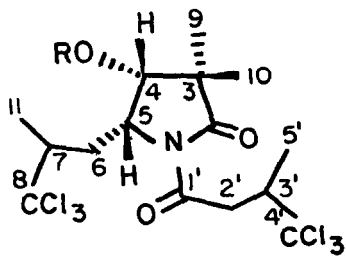
Dysidamide (2) was the major metabolite obtained from the chloroform fraction of the
ethanol extract of the dried sponge collected near Massawa in the Red Sea Repeated column
chromatography on silica gel (CHCl₃ pet. ether 1:1) afforded dysidamide (2) which
crystallized as fine white needles (m p 123-124°C, [α]_D²¹ - - 14.9° (c = 1, CHCl₃) The
molecular weight and the formula C₁₅H₂₁Cl₆NO₃ was established by chemical ionization mass
spectrometry and ¹³C-NMR

The IR spectrum (KBr) indicated the presence of hydroxyl (3540, 3460cm⁻¹), carbonyl
(1720cm⁻¹) and gem dimethyl (doublet at 1380cm⁻¹) groups The mass spectrum (CI-methane)
showed a cluster of peaks around m/z 476 (82%, MH⁺) with a pattern characteristic of
hexachlorinated compounds⁴ The major peaks at m/z 440 (100%, MH⁺-HCl), 404 (5%, MH⁺-2HCl),
358 (13%, MH⁺-CCl₃), 316 (7%, MH⁺-C₄H₆Cl₃), 288 (18%, MH⁺-C₅H₆Cl₃O) 252 (13%, 288-HCl) and
128 (60%, CH(OH)CH=N⁺HC(O)CMe₂) could all be accounted for by structure (2)

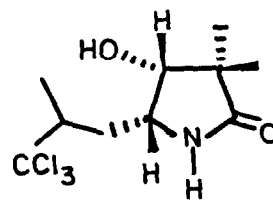
The ¹³C NMR spectrum showed two singlets at 106.97 and 105.86 ppm typical of the
trichloromethyl grouping^{3a} and two carbonyl signals at 179.84 and 172.39 ppm which agree



1



2 Dysidamide, R = H
4 R = Ac



3

with an N-acylated-2-pyrrolidinone system^{3a} The remainder of the ¹³C NMR spectrum consisted of resonances (Table 1) which accounted for the remaining 11 carbon atoms of the molecule Valuable information was obtained from the 360 MHz ¹H NMR spectra of dysidamide (2), its acetate⁵ and the hydrolysis product 3 (Table 1) The structure was substantiated both by a C-H correlation (Table 1) and a H-H COSY experiments⁶. The later experiment enabled the assignment of all the protons in the molecule The relative stereochemistry at C-4 and C-5 was determined from NOE measurements on the acetate derivative, 4 (9% enhancement between H-4 & 5)

TABLE 1 ¹H & ¹³C NMR Data and H-C Correlations of Compounds 2 & 3 in d₆-acetone

C-H [#]	2		3	
	δ _C	δ _H	δ _C	δ _H
1-H				6 80 brs (NH)
C2	179 84 s		179 69 s	
C3	46 55 s		44.20 s	
C4-H	72 50 d	4 24 dd(J=5 2, 7 6 Hz)	76.29 d	3 94 brd(J=5 1Hz)
		4 90 d(J=5 2 Hz, OH)		4 30 brs(OH)
C5-H	57 64 d	4 40 ddd(J=4 2, 5 9, 7 6 Hz)	51 54 d	3 77 ddd(J=10 2, 5 1, 3 6 Hz)
C6-H	33 70 t	2 30 ddd(J=4 2, 9 7, 14 1 Hz)	33.36 t	2 22 ddd(J=13 4, 10 2, 2.4 Hz)
		2 10 ddd(J=2 6, 5 9, 14.1 Hz)		1 60 ddd(J=13 4, 10 4, 3.6 Hz)
C7-H	54 27 d	3 07 ddq(J=2 6, 9 7, 6.4 Hz)	52 78 d	1 93 ddq(J=10 4, 2 4 6.4 Hz)
C8	106 97 s		106 43 s	
C9	19 50 q	1 14 s	17 60 q	0 96 s
C10	23 87 q	1.16 s	22 49 q	0.97 s
C11	16 65 q	1 37 d(J=6 4 Hz)	15.16 q	1.26 d(J=6 4 Hz)
C1'	172 39 s			
C2'-H	42 00 t	3 52 brd(J=16 9, 1 7 Hz)		
		3 20 dd(J=16 9, 9 5 Hz)		
C3'-H	51 34 d	3 23 ddq(J=1 7, 9 5, 6 3 Hz)		
C4'	105 86 s			
C5'-H	17 00 q	1 33 d(J=6 3 Hz)		

C-H Correlations C-No/H-No's 2/4, 9 & 10, 3/9, 10, 5 & OH, 4/6, 9, 10, 5/OH; 6/11, 7/5; 8/6, 11, 9/10, 4, 10/9, 11/6, 6, 1'/2', 2', 2'/5', 3'/2', 5', 4'/2', 5', 5'/2'

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- F W McLafferty, "Interpretation of Mass Spectra", 3rd ed, University Science Books, Mill Valley, California, 1980, p 277
- Compound 4. δ_H (CDCl₃, 360MHz) 5 19d(H-4, J=7 4), 4 57ddd(H-5, J=7.0, 7 4, 2 3), 3 61ddd(H-2', J=18 1, 2, 3), 3 32ddq(H-3', J=2 3, 9 4, 6 4), 3 17dd(H'-2', J=18 1, 9 4), 2 99ddq(H-7, J=2.6, 9.8, 6.4), 2 25ddd(H-6, J=14.5, 6.1, 2.3), 2.20s(AcO-4), 1.86ddd(H'-6, J=14.5, 9.8, 4.0) 1 39d(Me-5', J=6 4), 1.32d(Me-11, J=6.4), 1 28s(Me-10), 1.24s(Me-9).
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